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jan 21 2023 what is the correct way to prepare a 1 molar solution of KNO_3 given the choices a add one gram of KNO_3 to 99 grams of water b combine 101.1 g of KNO_3 with 1 liter of water c combine 101.1 grams of KNO_3 with 1 kilogram of water d add 101.1 grams of KNO_3 to a small volume in water in a 1 liter volumetric flask dissolve the KNO_3 and then fill click here to get an answer to your question calculate the boiling point elevation for a solution prepared by adding 10 gram of calcium chloride to 200 gm of water kb for water 0.512 K kg mol^{-1} molar mass of CaCl_2 mol 1 what do you do 42 list two methods that could be used to determine the end 5 if 10ml of 2.0M solution is diluted to 125ml with distilled water to prepare solution a 10ml of solution a is put into a measuring cylinder and diluted to 125ml of distilled water to prepare b what are the molarities of the resulting solutions a and b respectively 1 day ago water was used to prepare all the required solutions 0.1 mol/l of $\text{Zn}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$ solution was prepared and the solution was stirred for 10 minutes to prepare ZnO nanoparticles this solution was mixed with 5 ml of ammonia solution to prepare a precipitating agent after stirring this solution for 20 minutes it was irradiated in the

molar concentration also called molarity amount concentration or substance concentration is a measure of the concentration of a chemical species in particular of a solute in a solution in terms of amount of substance per unit volume of solution in chemistry the most commonly used unit for molarity is the number of moles per liter having the unit symbol mol l or mol dm^3 in si unit glucose molar mass is $180.16 \text{ g mol}^{-1}$ is a simple soluble sugar glucose solutions are used to treat patients with low blood sugar suppose you prepare a glucose solution using the described procedure step 1 dissolve 23.16 g of glucose in enough water to make 500.0 ml of solution step 2 transfer 15.2

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